

A NOVEL PROCEDURE FOR THE PREPARATION OF 3-SUBSTITUTED 4-HYDROXYPYRAZOLE-5-CARBOXYLIC ACIDS AS AN APPROACH TO PYRAZOMYCIN

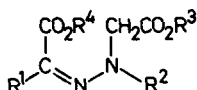
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As a part of our program directed toward the synthesis of C-nucleosides we have devised a general method for the synthesis of 3-substituted 4-hydroxypyrazole-5-carboxylic acids as a route to pyrazomycin¹.



Ia R¹ = Me; R² = H; R³ = R⁴ = Et

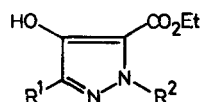
Ib R¹ = R² = R³ = Me; R⁴ = Et

Ic R¹ = Ph; R² = H; R³ = Et; R⁴ = Me

Id R¹ = R³ = Me; R² = Ph; R⁴ = Et

Ie R¹ = R² = Ph; R³ = R⁴ = Me

If R¹ = t-Bu; R² = H; R³ = Et; R⁴ = Me



IIa R¹ = Me; R² = H

IIb R¹ = R² = Me

IIc R¹ = Ph; R² = H

IId R¹ = Me; R² = Ph

IIe R¹ = R² = Ph

IIf R¹ = t-Bu; R² = H

We have found that α -keto ester 2-carboalkoxymethylhydrazones undergo the Dieckmann reaction² to give 4-hydroxypyrazole-5-carboxylic acids. (In an analogous manner, some derivatives of β -hydroxypyrrole³ and β -hydroxythiophene⁴ have been prepared.) The starting hydrazones Ia-If were obtained by treatment of appropriate methyl or ethyl α -ketocarboxylic acid esters with ethyl hydrazinoacetate (preparation of Ia, Ic and If), with (1-methylhydrazino)-acetic acid or with (1-phenylhydrazino)acetic acid, followed by esterification with diazomethane (preparation of Ib and Id, Ie, resp.), as chromatographically homogenous oils. The Dieckmann cyclisation of Ia-If into 4-hydroxypyrazole-5-carboxylic acid esters IIa-IIf was carried out by refluxing 0.01 mole of

the starting hydrazones in 40 ml 0.5 N sodium ethoxide in ethanol for two hours. After neutralisation of the reaction mixture, the product was isolated on a column of silicagel. UV spectra of IIa-IIf exhibit a characteristic bathochromic shift due to formation of the enolate ion. The IR spectra of IIa-IIf

TABLE

Physical Properties of Ethyl 3-Substituted 4-Hydroxypyrazole-5-carboxylates

Comp.	Yield %	Melting point °C (solvent)	λ max in nm		(log ϵ)		IR (CHCl ₃)		ν in cm ⁻¹	
			0.1 N HCl	0.1 N NaOH			(C=O) ^a	(C=O) ^b	(OH)	(NH)
IIa	50	174-175 water	224 (3.87) 273 (3.75)	237 (3.83) 317 (3.87)	1692	1722	3539	3448		
IIb	52	34-35 ether-p.ether ^c	235 (4.01) 279 (3.79)	242 (3.86) 323 (3.87)	1676	1725	3521 3400			
IIc	39	171-172 ether-p.ether ^c	218 (4.22) ^d 237 (4.20) 281 (3.90)	228 (4.15) 333 (3.95)	1693	1720i	3440 3370	3390 3100		
IIId	45	oil distilled at 120 (0.05 mm)	236 (3.97) 288 (3.87)	251 (3.83) 331 (4.06)	1671	1721 1731	3505 3380			
IIe	85	95-97 p.ether ^c	225 (4.23) ^d 240 (4.19) 304 (3.98)	228 (4.23) 347 (4.10)	1673	1720 1737	3480 3330			
IIIf	41	124-125 10% aq. acetic acid	223 (3.84) 272 (3.70)	237 (3.80) 320 (3.91)	1687	1721i	3525	3450 3140		

^aEnolic β -keto ester. ^bNon-enolic β -keto ester. ^cPetroleum ether b.p. 60-70°C.

^dMeasured in 0.1 N HCl in 10% ethanol.

provide evidence of keto-enol equilibria in chloroform solution, with prevailing enol form. IIa-IIf give a positive ferric chloride test.

All compounds described gave satisfactory elemental analyses and their mass spectra fragmentation patterns are compatible with the assigned structures.

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